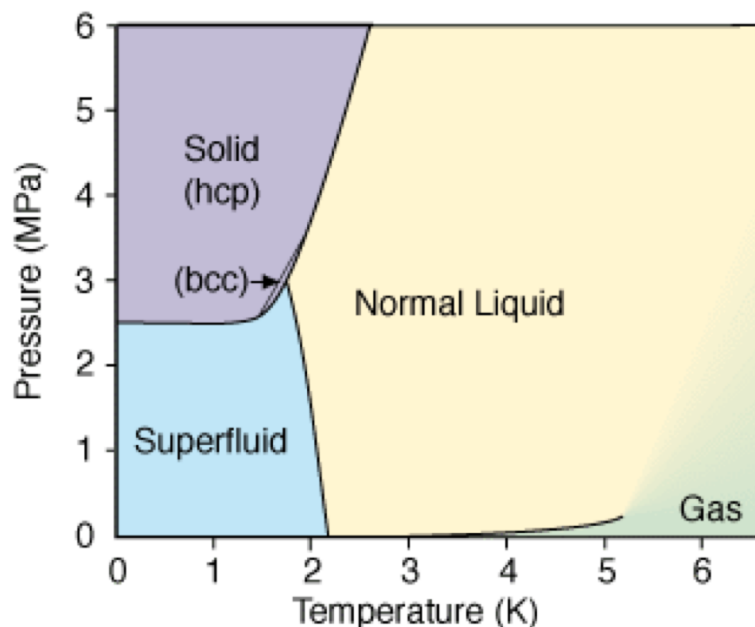
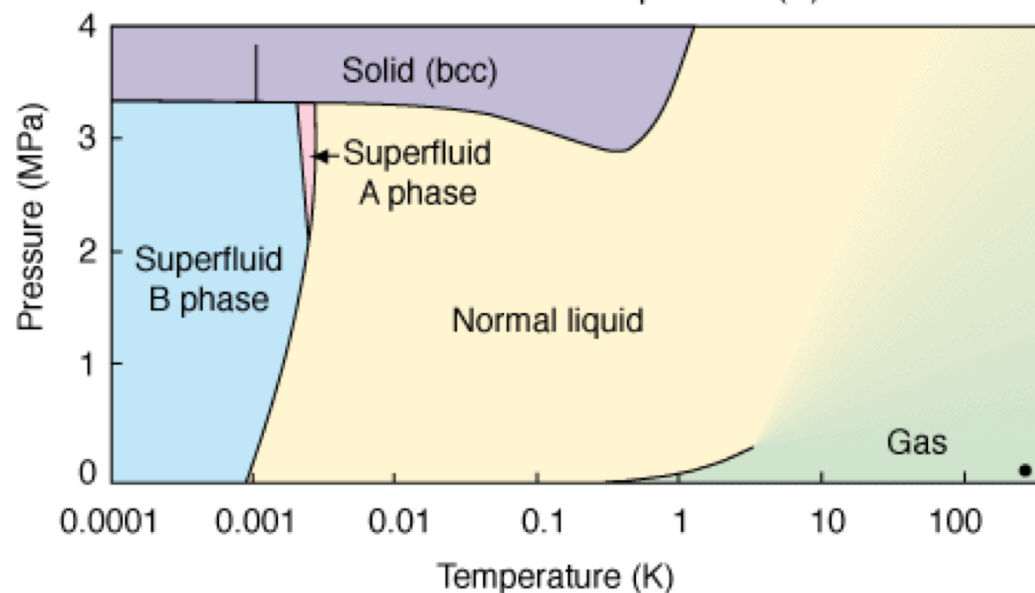


Superfluidità nell'Elio



^4He
si comporta da
Bosone e a basse T ha
la **BEC** (Bose-Einstein
Condensation)

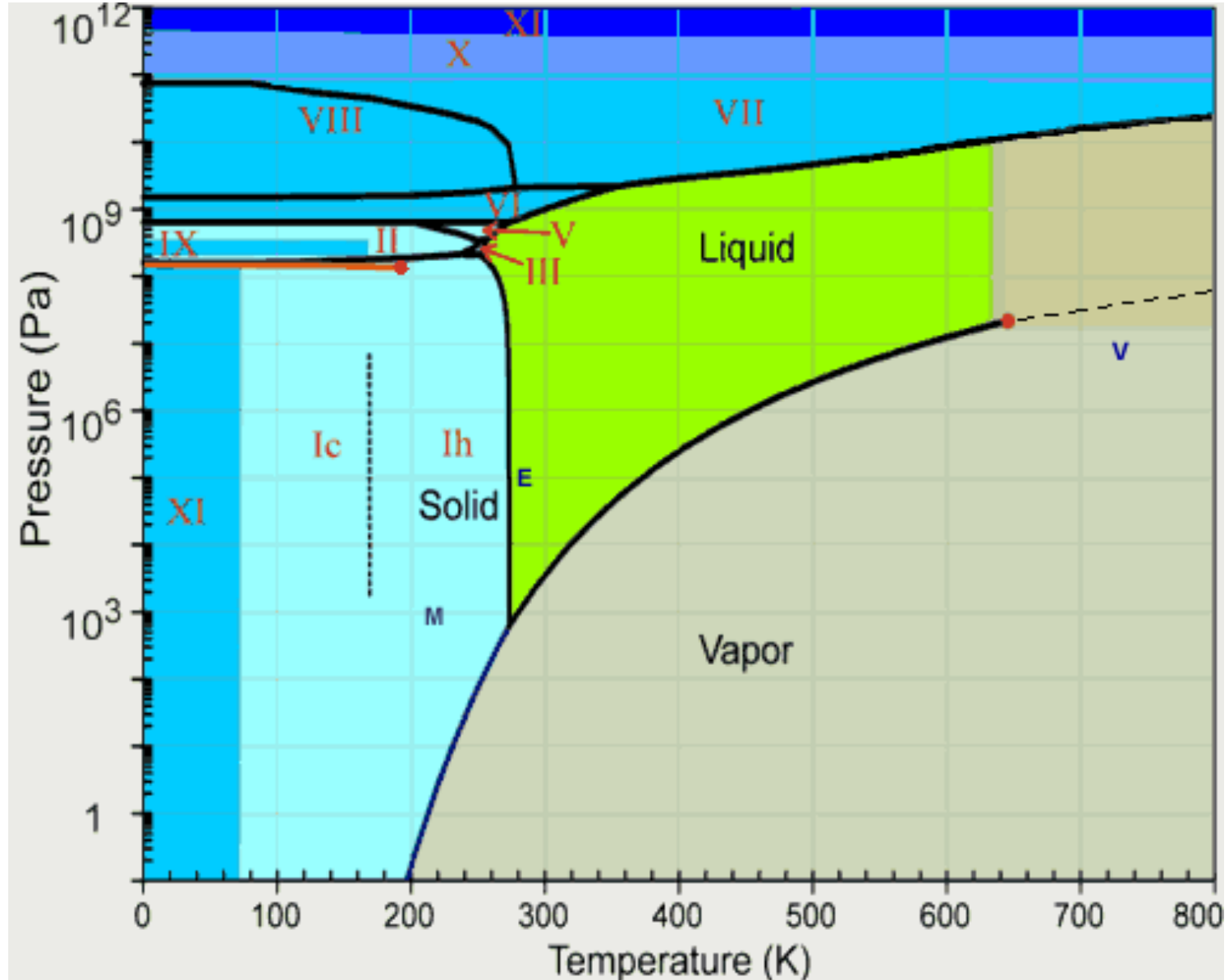


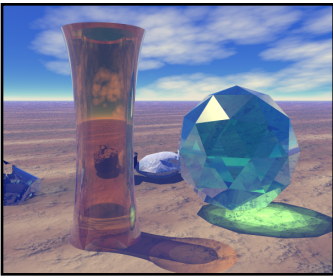
^3He
Sarebbe un Fermione
ma a basse T forma
Coppie di Cooper che si
comportano da **Bosoni**)

Periodic Table of the Elements

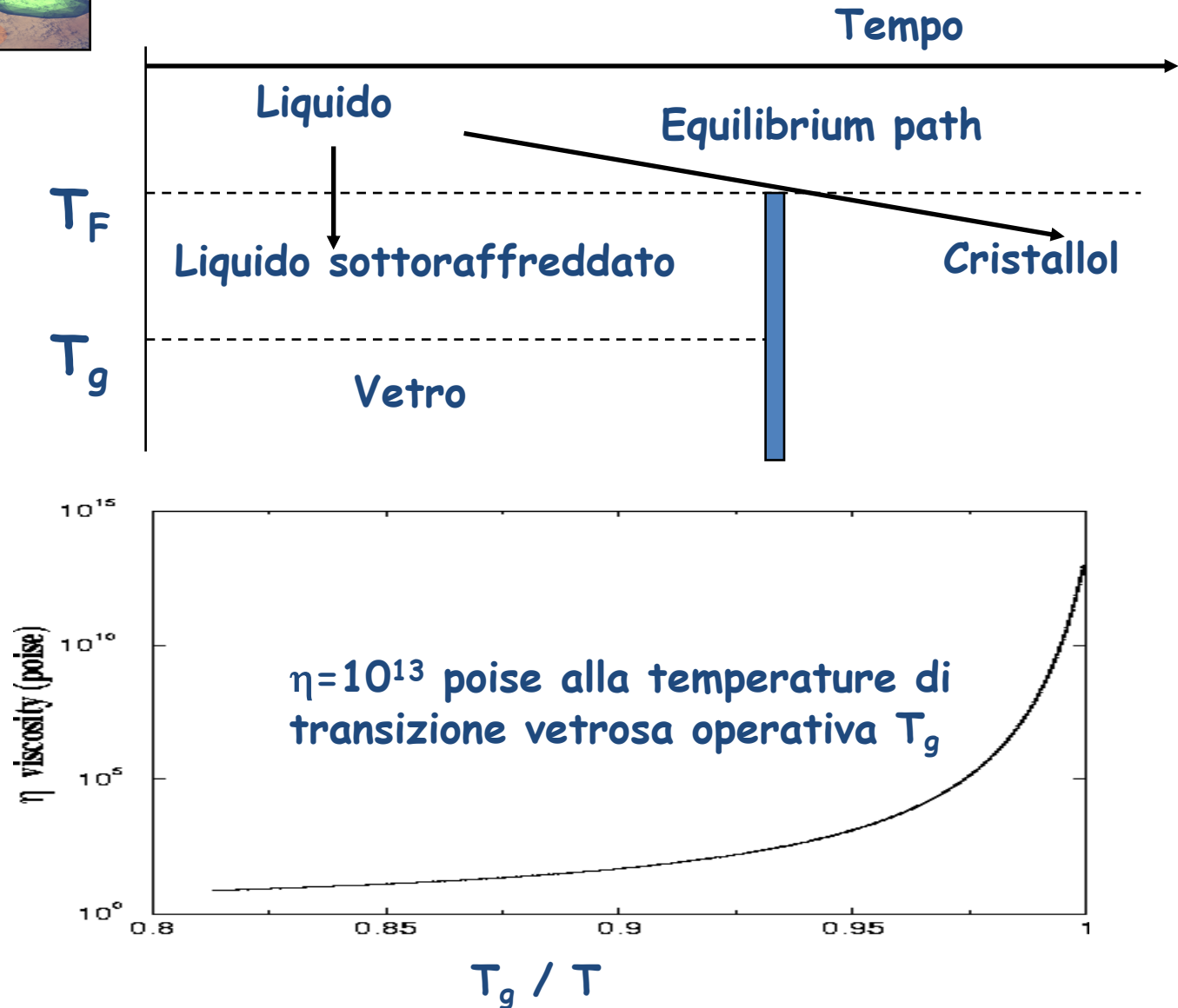
		Groups →																																																																													
Periods ↓		1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18																																																												
		Alkali Metals	Alkali Earth Metals	Transition Metals	Rare Earth Metals	Other Metals	Non Metals	Halogens	Noble Gases																																																																						
1	1	H																	2	He																																																											
2	3	Li	4	Be										5	B	6	C	7	N	8	O	9	F	10	Ne																																																						
3	11	Na	12	Mg										13	Al	14	Si	15	P	16	S	17	Cl	18	Ar																																																						
4	19	K	20	Ca	21	Sc	22	Ti	23	Y	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr																																											
5	37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe																																											
6	55	Cs	56	Ba	*La	72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	81	Tl	82	Pb	83	Bi	84	Po	85	At	86	Rn																																												
7	87	Fr	88	Ra	+Ac	104	Rf	105	Ha	106	106	107	107	108	108	109	109	110	110																																																												
		Electron Surplus ← Reducing Elements Oxidizing Elements → Electron Deficit																																																																													
		<table><tr><td>58</td><td>59</td><td>60</td><td>61</td><td>62</td><td>63</td><td>64</td><td>65</td><td>66</td><td>67</td><td>68</td><td>69</td><td>70</td><td>71</td><td></td></tr><tr><td>Ce</td><td>Pr</td><td>Nd</td><td>Pm</td><td>Sm</td><td>Eu</td><td>Gd</td><td>Tb</td><td>Dy</td><td>Ho</td><td>Er</td><td>Tm</td><td>Yb</td><td>Lu</td><td>Lanthanides</td></tr><tr><td>90</td><td>91</td><td>92</td><td>93</td><td>94</td><td>95</td><td>96</td><td>97</td><td>98</td><td>99</td><td>100</td><td>101</td><td>102</td><td>103</td><td></td></tr><tr><td>Th</td><td>Pa</td><td>U</td><td>Np</td><td>Pu</td><td>Am</td><td>Cm</td><td>Bk</td><td>Cf</td><td>Es</td><td>Fm</td><td>Md</td><td>No</td><td>Lr</td><td>Actinides</td></tr></table>														58	59	60	61	62	63	64	65	66	67	68	69	70	71		Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Lanthanides	90	91	92	93	94	95	96	97	98	99	100	101	102	103		Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	Actinides				
58	59	60	61	62	63	64	65	66	67	68	69	70	71																																																																		
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Lanthanides																																																																	
90	91	92	93	94	95	96	97	98	99	100	101	102	103																																																																		
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	Actinides																																																																	

Diagramma di fase dell'acqua

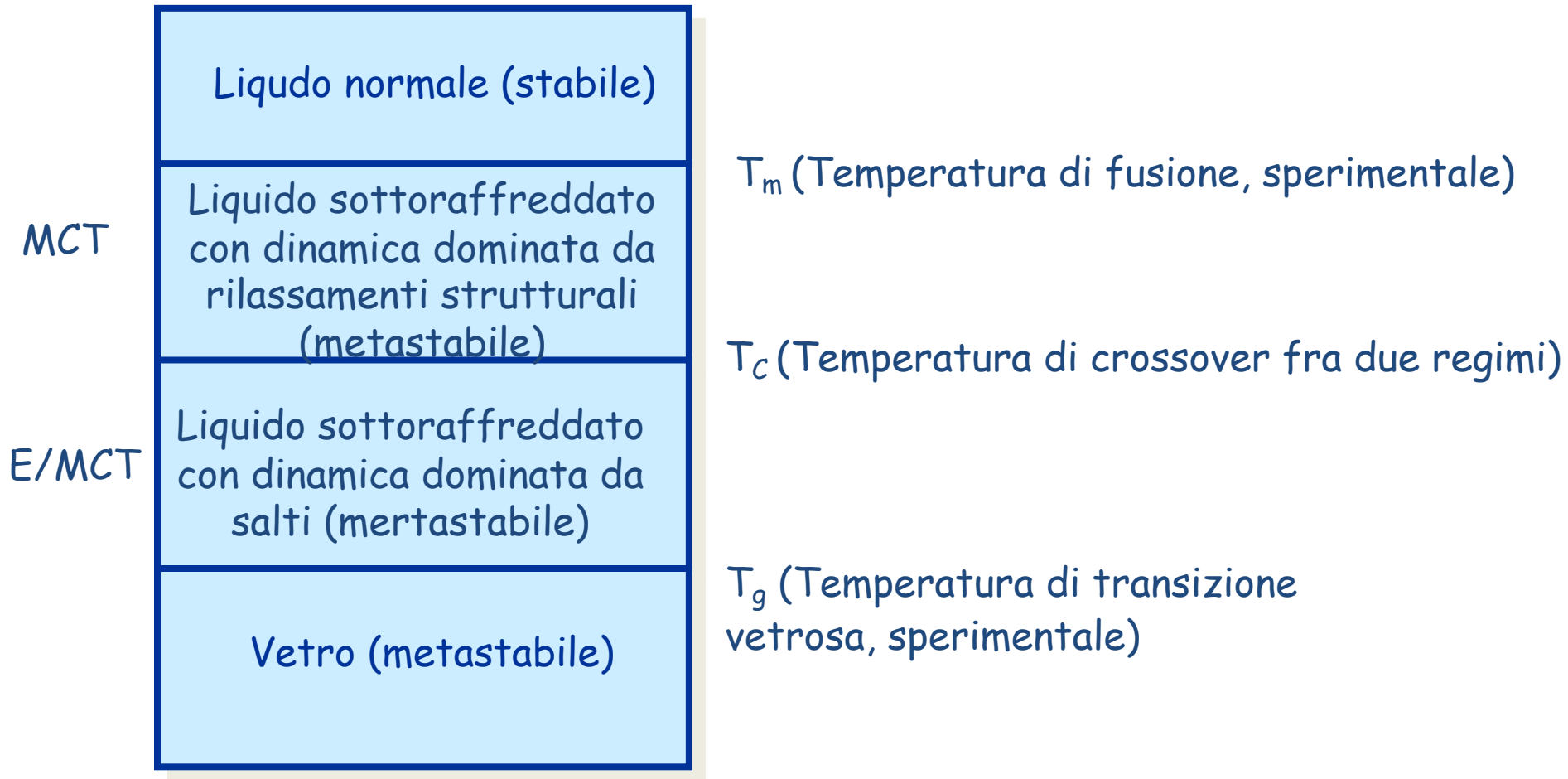




Liquidi sottoraffreddati e transizione vetrosa

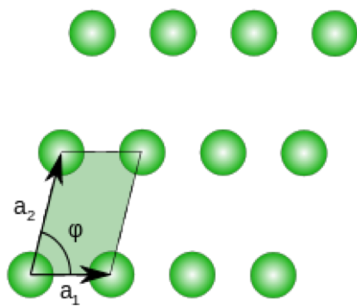


Transizione vetrosa



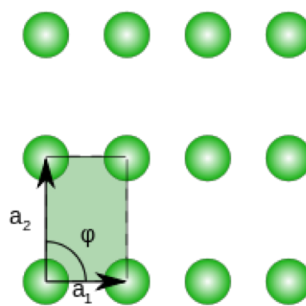
[Complex dynamics of glass forming liquids: a mode coupling theory, W. Götze, Oxford (2009)]

I cinque diversi reticoli di Bravais in due dimensioni: obliquo, rettangolare, rettangolare centrato, esagonale e quadrato



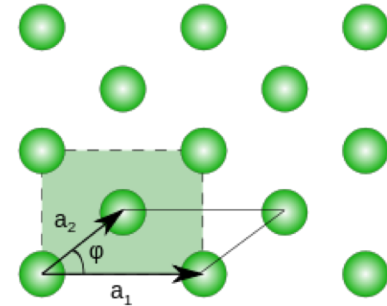
$$|a_1| \neq |a_2|, \varphi \neq 90^\circ$$

1



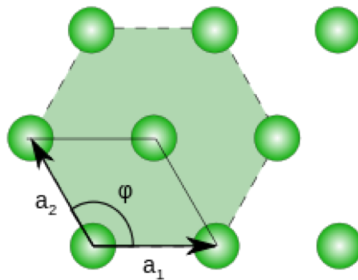
$$|a_1| \neq |a_2|, \varphi = 90^\circ$$

2



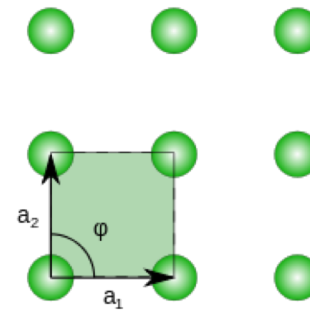
$$|a_1| \neq |a_2|, \varphi \neq 90^\circ$$

3



$$|a_1| = |a_2|, \varphi = 120^\circ$$

4



$$|a_1| = |a_2|, \varphi = 90^\circ$$

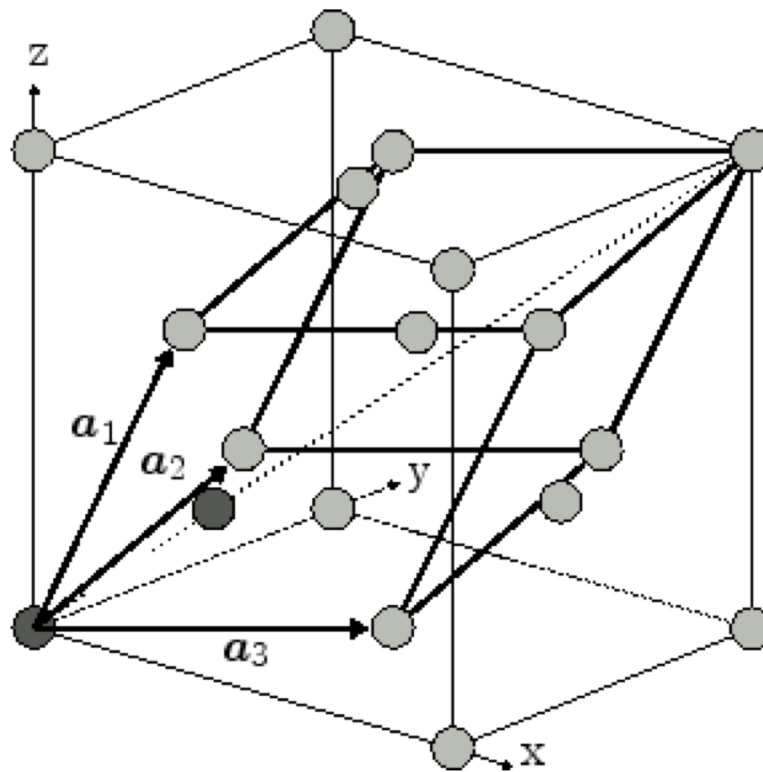
5

Reticoli di Bravais 3DIM

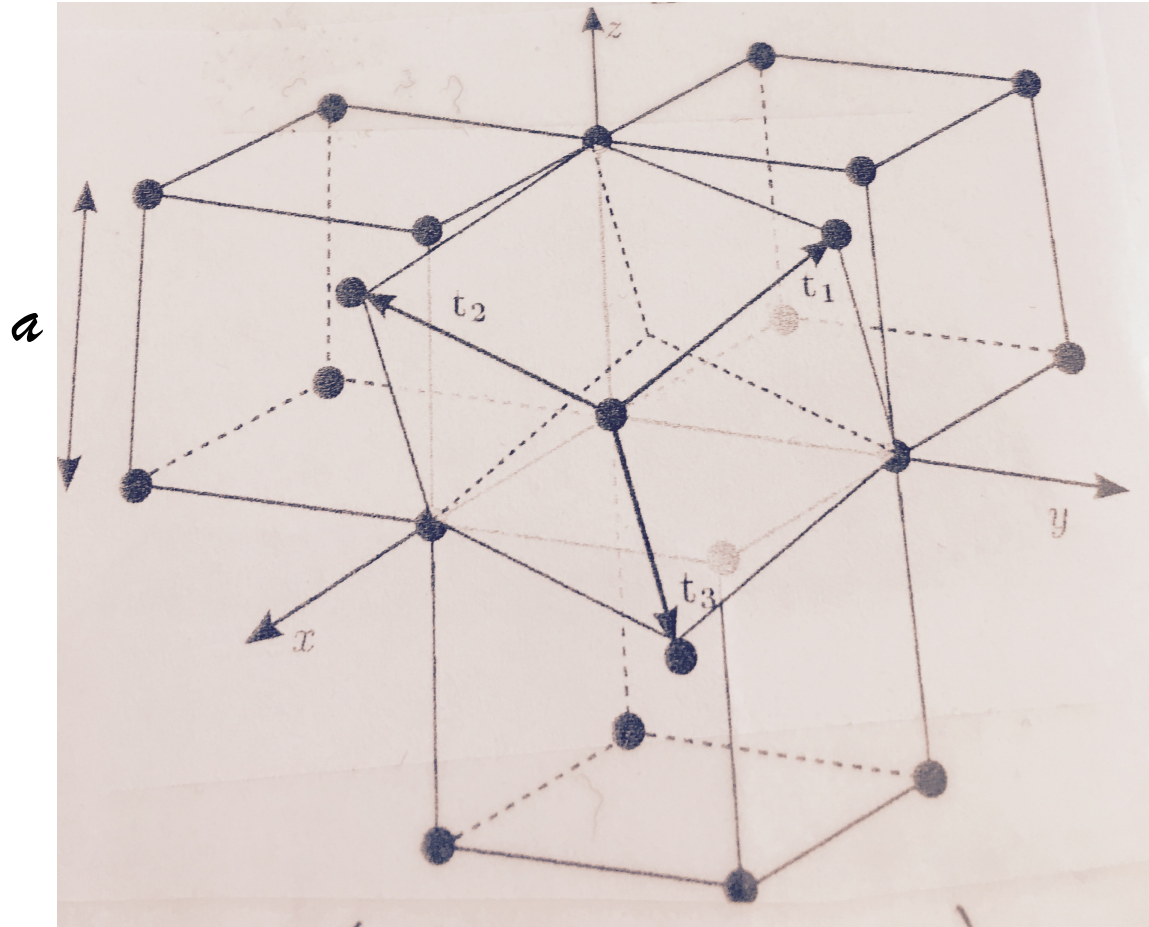
Crystal system	Bravais lattices			
	primitive	base-centered	body-centered	face-centered
Triclinic $a \neq b \neq c$ $\alpha \neq \beta \neq \gamma$				
Monoclinic $a \neq b \neq c$ $\alpha = \gamma = \frac{\pi}{2} \neq \beta$				
Orthorhombic $a \neq b \neq c$ $\alpha = \beta = \gamma = \frac{\pi}{2}$				
Trigonal $a = b = c$ $\alpha = \beta = \gamma \neq \frac{\pi}{2}$				
Tetragonal $a = b \neq c$ $\alpha = \beta = \gamma = \frac{\pi}{2}$				
Hexagonal $a = b \neq c$ $\alpha = \beta = \frac{\pi}{2}$ $\gamma = \frac{2\pi}{3}$				
Cubic $a = b = c$ $\alpha = \beta = \gamma = \frac{\pi}{2}$				

Fig. 3 Classification of the fourteen Bravais lattices into the seven crystal systems. The conventional parallelepiped cells are shown.

Vettori di base primitivi del reticolo cubico a face centrate (face centered cubic, fcc) che formano la cella primitiva. La cella primitiva in figura ha anche una base di due atomi che sono segnati in colore più scuro.

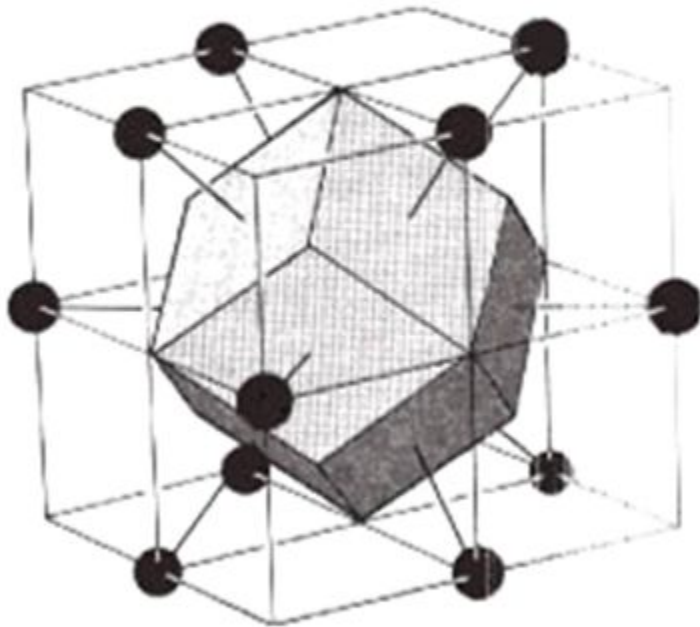


Cella primitiva del cristallo cubico a face cetrato (Body Centered Cubic) BCC



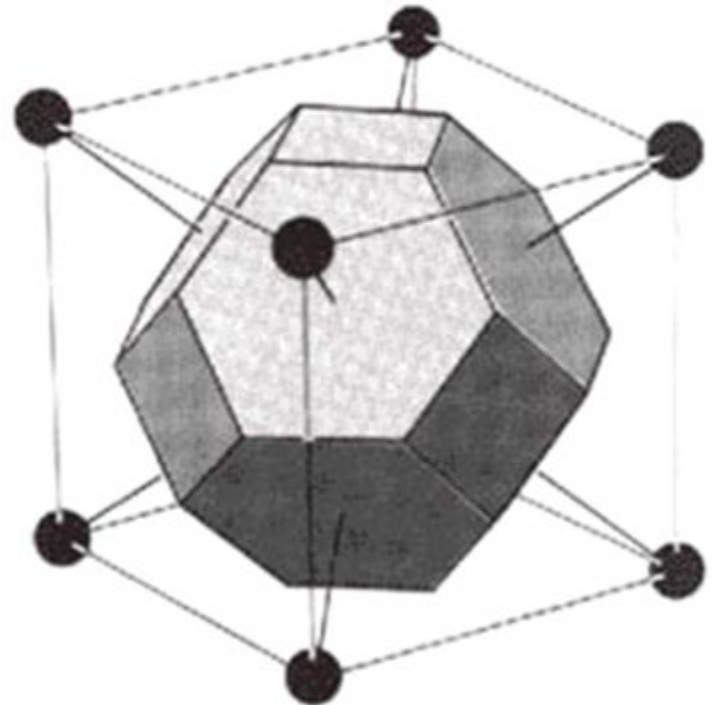
Wigner-Seitz Cell - 3D

Dodecaedro rombico: ha 12 facce a forma di rombo

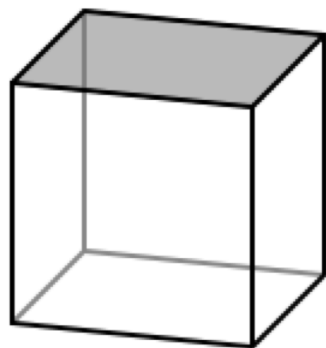


f.c.c Wigner-Seitz cell

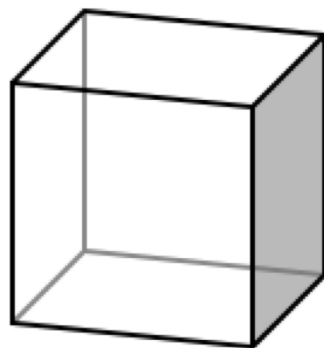
Ottaedro troncato (uno dei tredici poliedri archimedei ottenuto tagliando le cuspidi all'ottaedro regolare che e' un solido platonico)



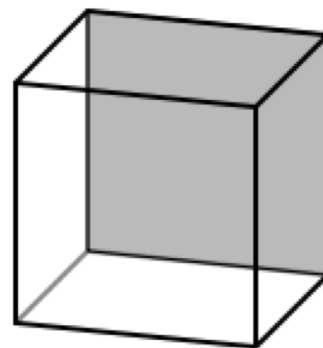
b.c.c Wigner-Seitz cell



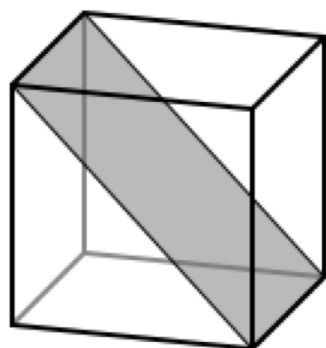
(001)



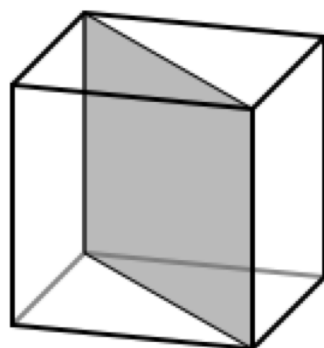
(100)



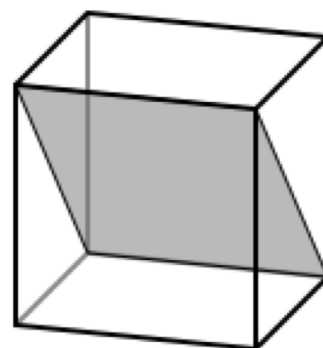
(010)



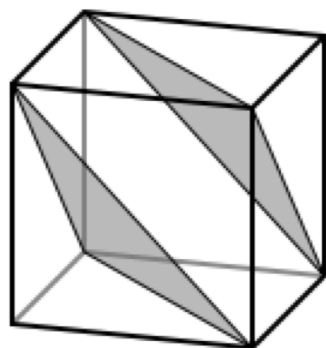
(101)



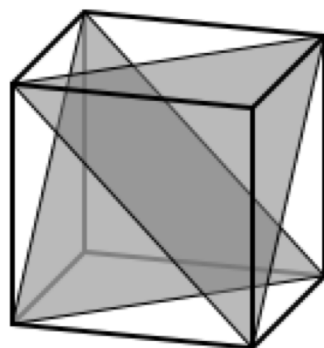
(110)



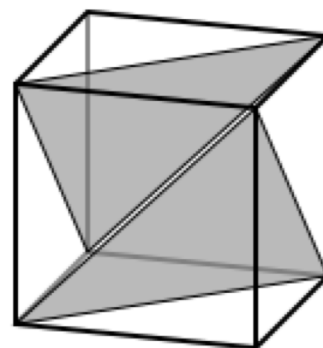
(011)



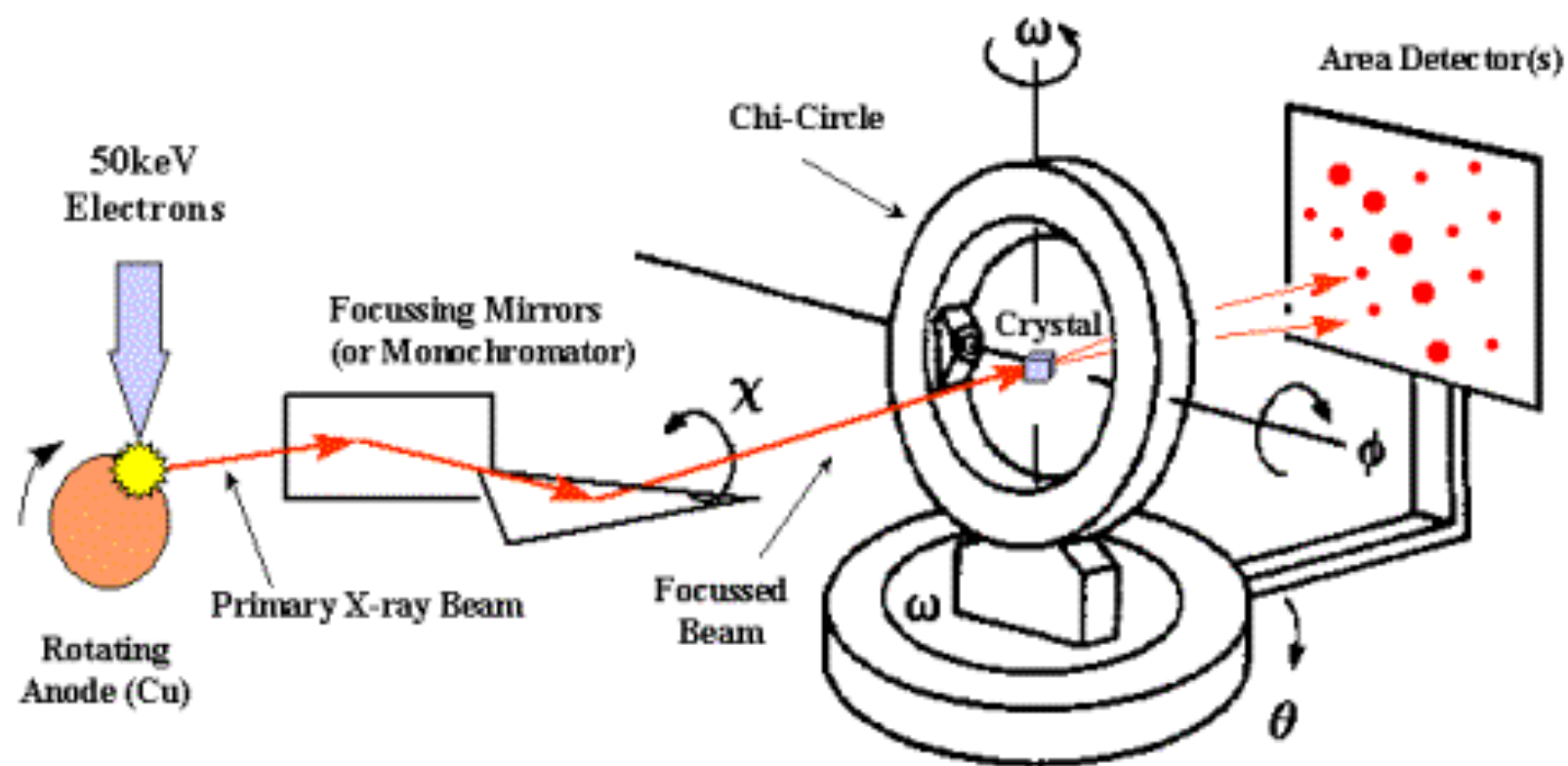
(111)



$(1\bar{1}1)$



$(\bar{1}11)$



4-Circle Goniometer (Eulerian or Kappa Geometry)

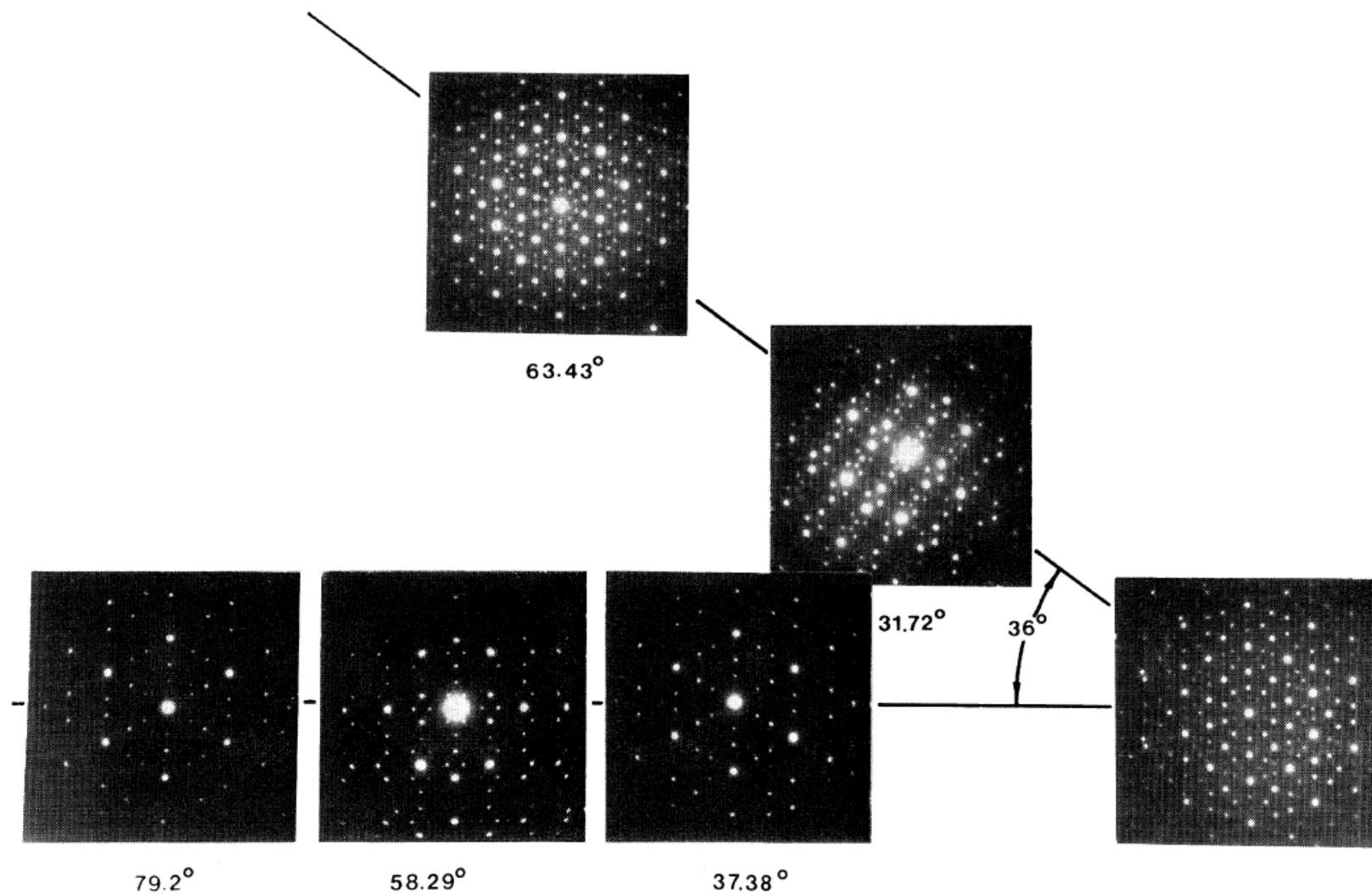


FIG. 2. Selected-area electron diffraction patterns taken from a single grain of the icosahedral phase. Rotations match those in Fig. 1.

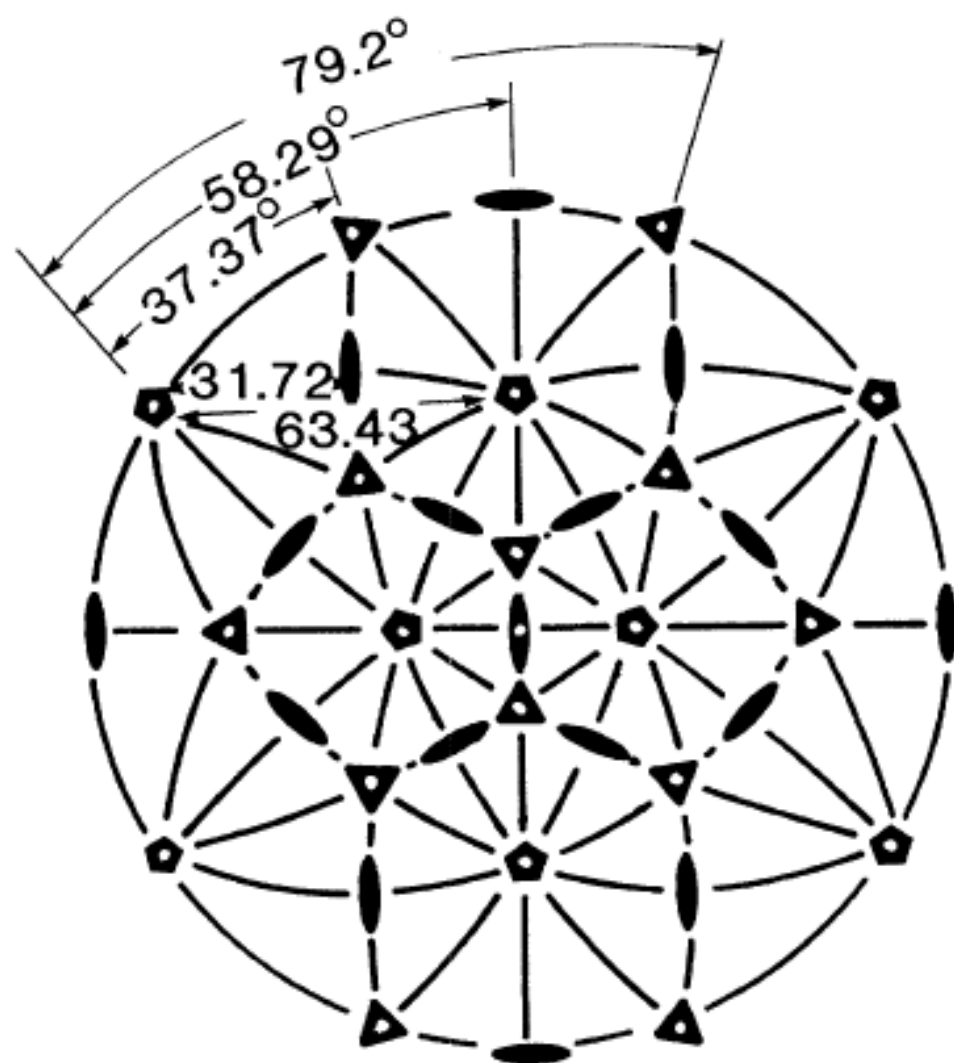
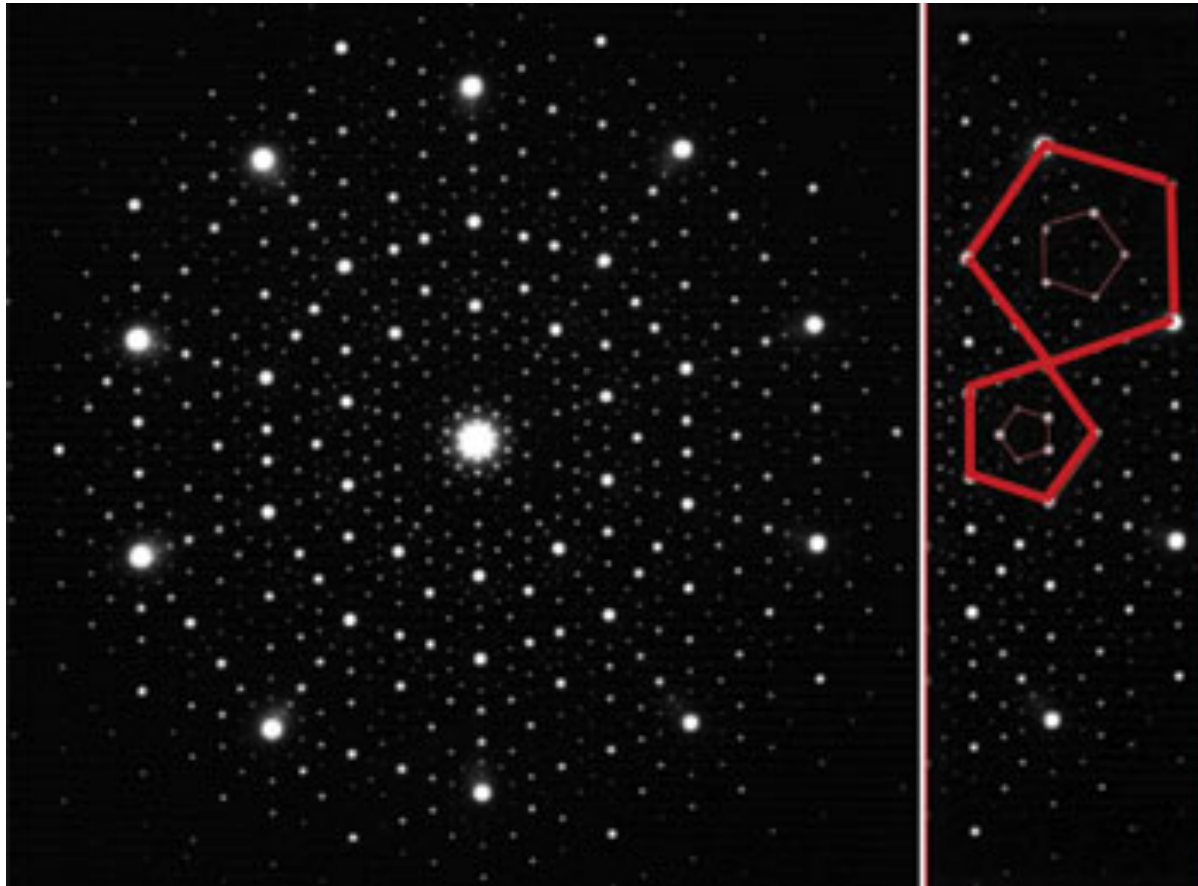
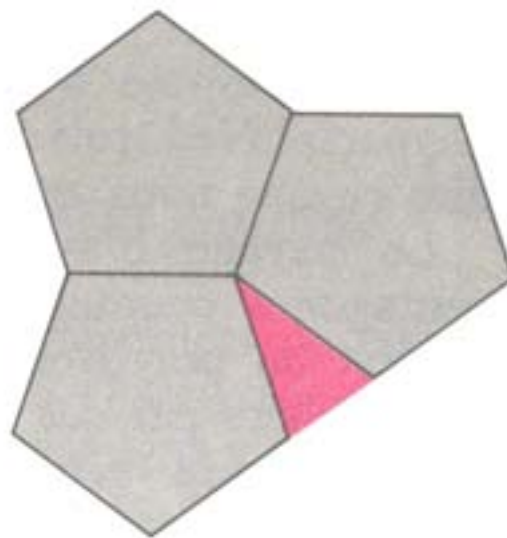
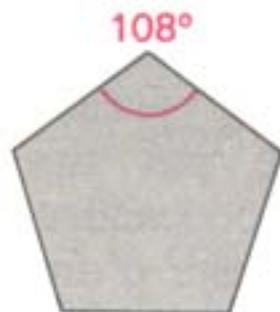


FIG. 1. Stereographic projection of the symmetry elements of the icosahedral group $m\bar{3}5$.



The electron diffraction pattern from an icosahedral quasicrystal contains perfect pentagons. The 2011 Nobel prize in chemistry was awarded to Daniel Shechtman of the Technion - Israel Institute of Technology, Haifa, for the discovery of quasicrystals. Quasicrystals exhibit long-range order but contain atoms packed in a pattern that cannot be repeated - a concept that broke the well established rules of crystalline behaviour at the time of their discovery in 1984.

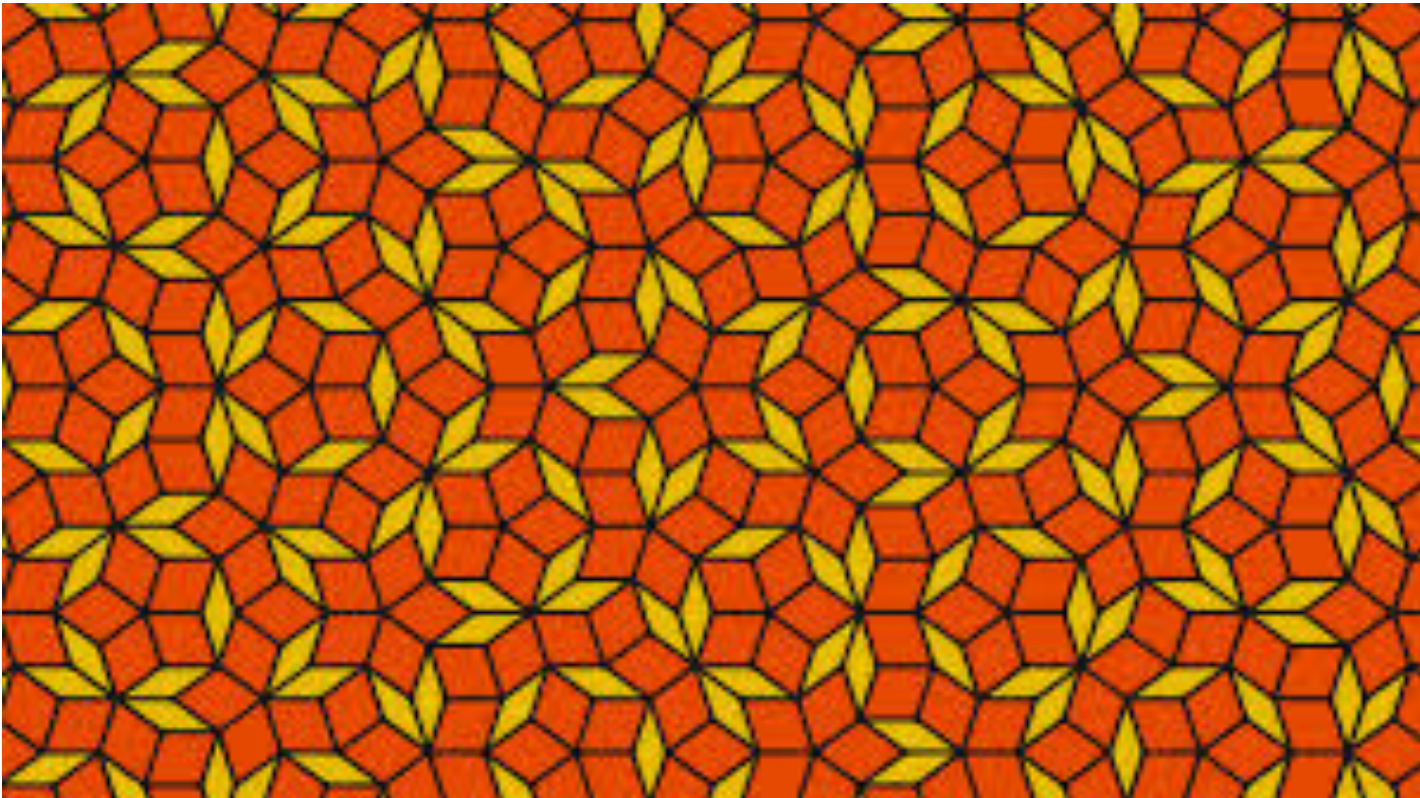
Frustrazione Geometrica del poentagono



2DIM Penrose Tiling

L'ordine orientazionale si traduce nel fatto che i rombi non possono avere che 10 orientazioni possibili.

Si noti che la struttura quasicristallina 2Dim e' contruita con due tipi di rombi come celle elementari



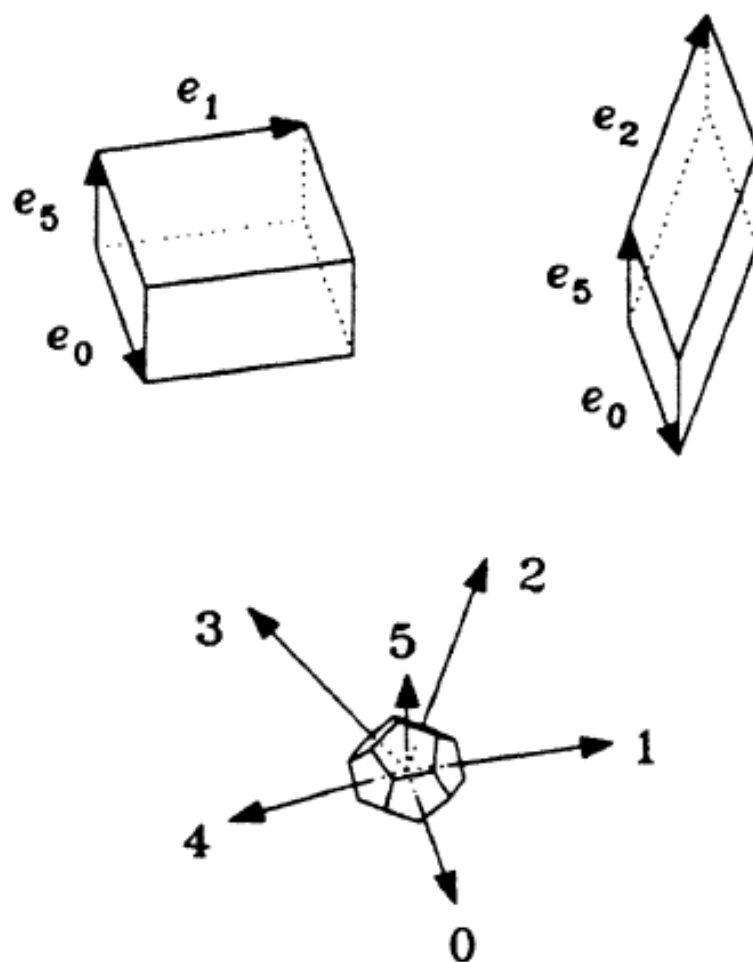
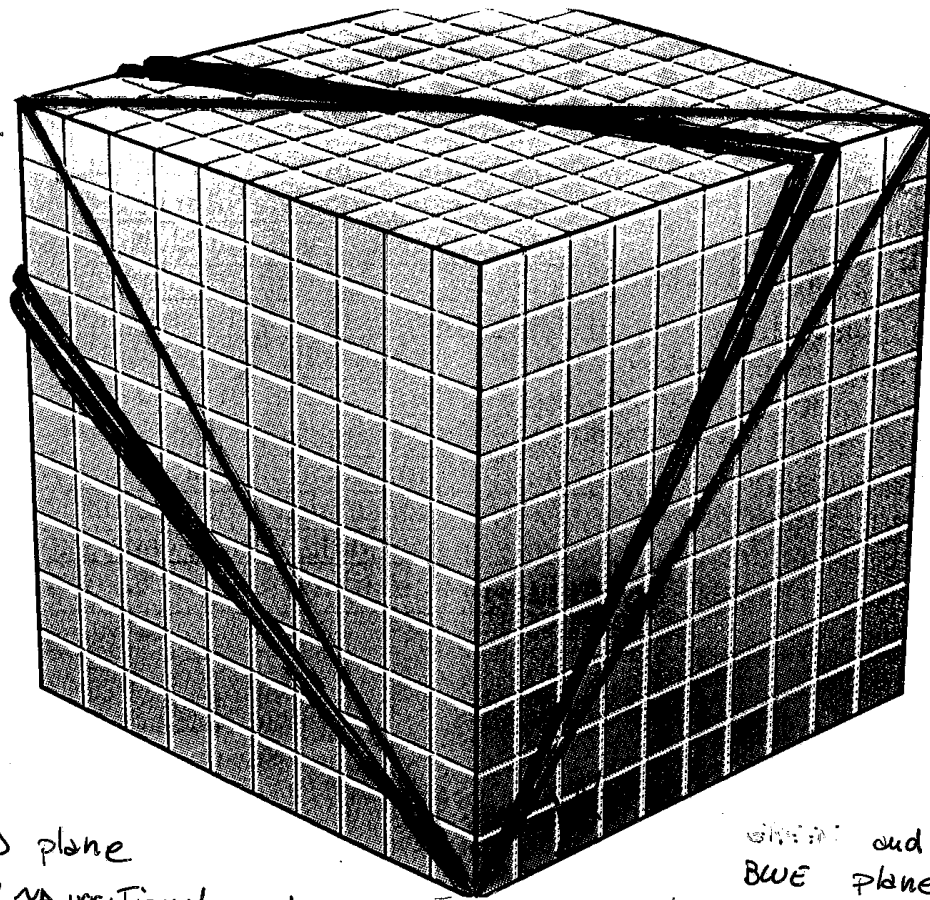


FIG. 8. Two types of rhombohedra that arise naturally in icosahedral quasicrystal packings and the star of vectors chosen for the analysis of the vertex model. The prolate rhombohedron (left) is based on a triplet of the form $(\mathbf{e}_5, \mathbf{e}_0, \mathbf{e}_1)$, the oblate on a triplet of the form $(\mathbf{e}_5, \mathbf{e}_0, \mathbf{e}_2)$.



RED plane
 $t/\alpha \sim \text{irrational number}$

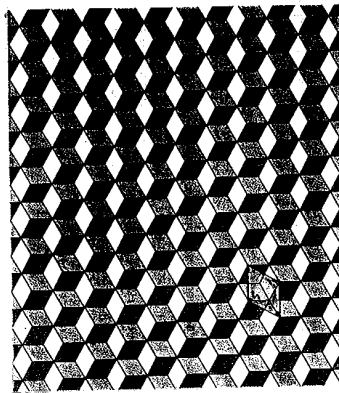
BLUE planes
 $t/\alpha \sim \text{rational number}$

- WHEN A 3-DIM CUBIC LATTICE IS CUT BY A PLANE EITHER PERIODIC OR QUASIPERIODIC 2-DIM LATTICES CAN BE OBTAINED.

THE INCINATION OF THE PLANE WITH RESPECT TO THE CUBIC LATTICE DETERMINES THE CHARACTERISTICS OF THE RESULTING 2-DIM LATTICE.

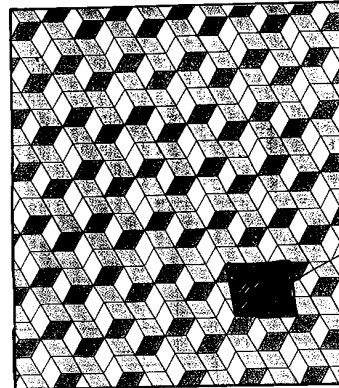
THE PLANES OF THE
PREVIOUS FIGURE INTERSECT
THE 3-DIM LATTICE SO THAT
THEY FORM:

- GREEN PLANE
Periodic lattice



unit cell

- BLUE PLANE
Periodic lattice



unit cell

- RED PLANE
Quasiperiodic lattice
NO SINGLE UNIT
CELL CAN GENERATE
THIS LATTICE

