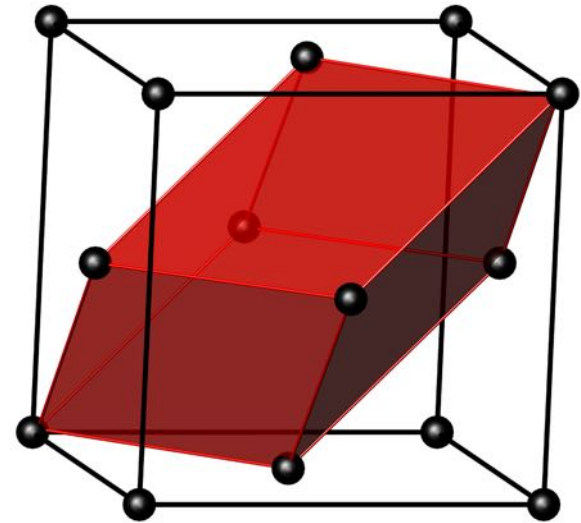
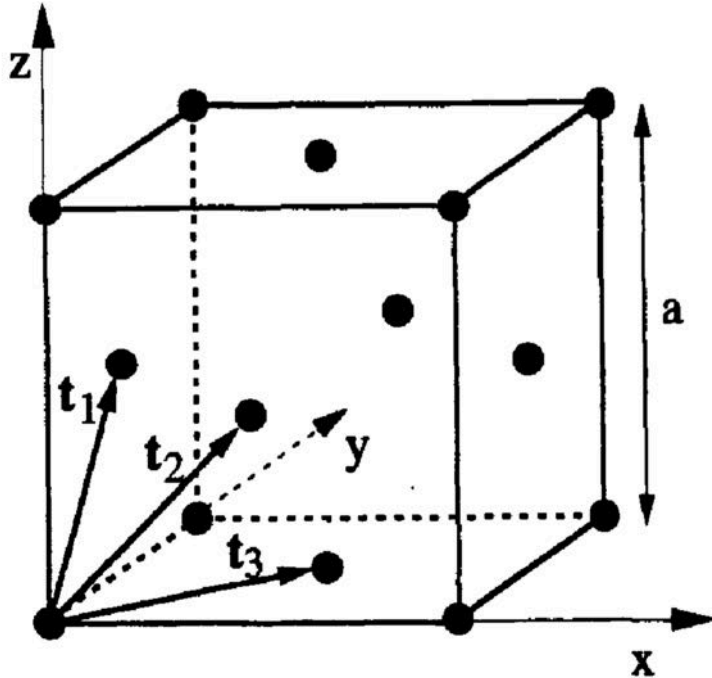


Reticoli Cristallini

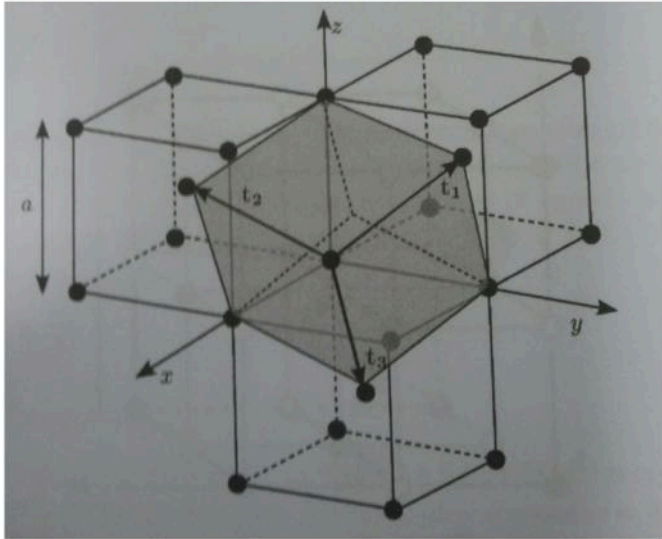
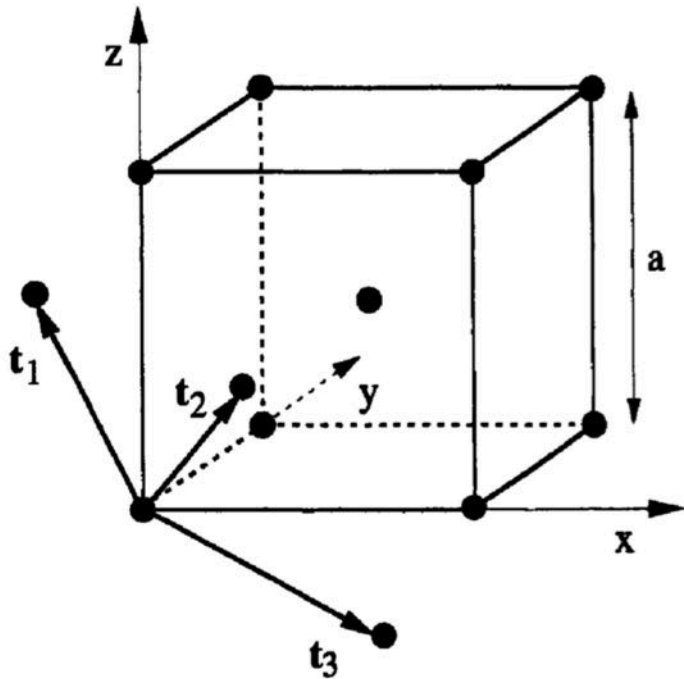
Cubico a facce centrate - fcc



- Solidi di gas rari (Ne, Ar, Kr, Xe)
- Vari metalli (Ag, Al, Au, Cu, Pd, Pt)
- Qualche elemento delle Terre Rare

Crystal	a (Å)	Crystal	a (Å)	Crystal	a (Å)	Crystal	a (Å)	Crystal	a (Å)
Ar	5.26	Au	4.08	Cu	3.61	Ni	3.52	Pt	3.92
Ag	4.09	Ca	5.58	Kr	5.72	Pb	4.95	Sr	6.08
Al	4.05	β -Co	3.55	Ne	4.43	Pd	3.89	Xe	6.2

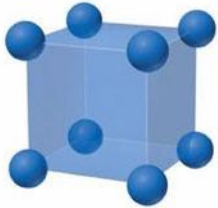
Cubico a corpo centrato - bcc



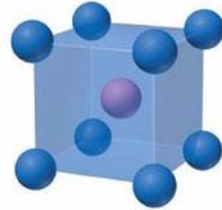
- Metalli alcalini (Li, Na, K, Rb, Cs)
- Altri metalli (Cr, Mo, W, Fe, Ba)

Crystal	a (Å)	Crystal	a (Å)	Crystal	a (Å)	Crystal	a (Å)
Ba	5.26	Fe	4.08	Mo	3.61	Rb	3.52
Cr	4.09	K	5.58	Na	5.72	Ta	4.95
Cs	4.05	Li	3.55	Nb	4.43	V	3.92
W	6.08						

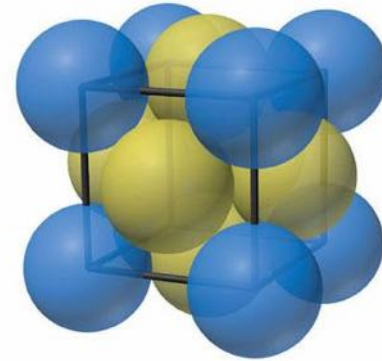
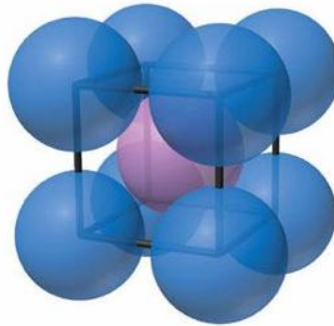
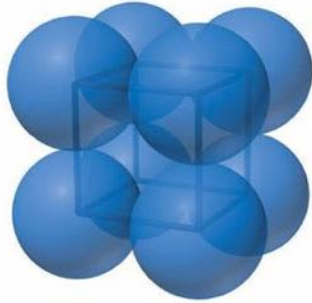
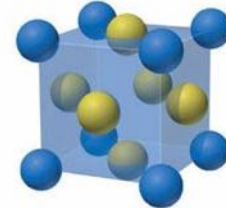
Simple cubic



Body-centered cubic



Face-centered cubic



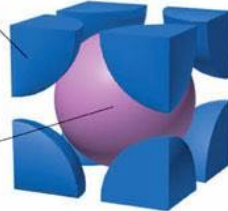
$\frac{1}{8}$ atom
at 8 corners



Atoms/unit cell = $\frac{1}{8} \times 8 = 1$

$\frac{1}{8}$ atom
at 8 corners

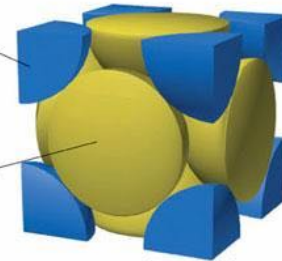
1 atom
at center



Atoms/unit cell = $(\frac{1}{8} \times 8) + 1 = 2$

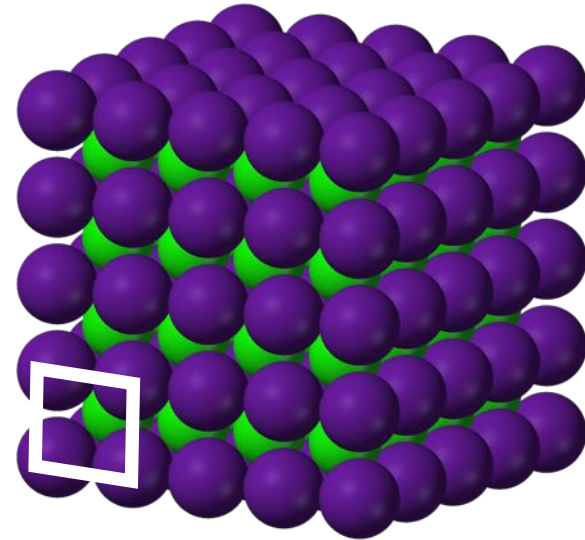
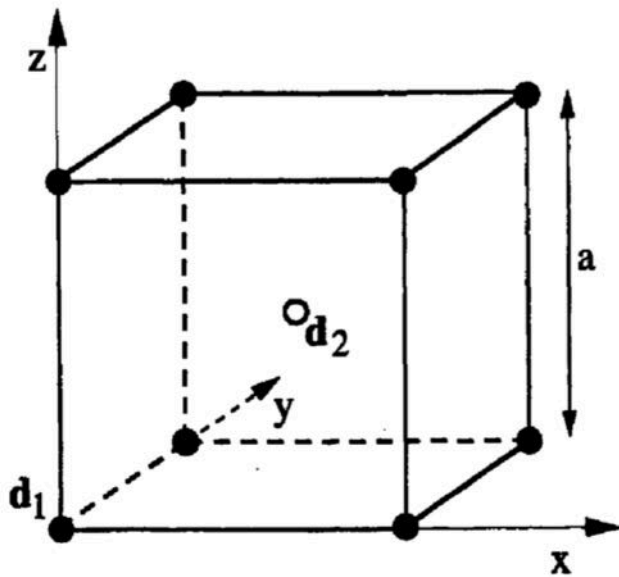
$\frac{1}{8}$ atom
at 8 corners

$\frac{1}{2}$ atom
at 6 faces



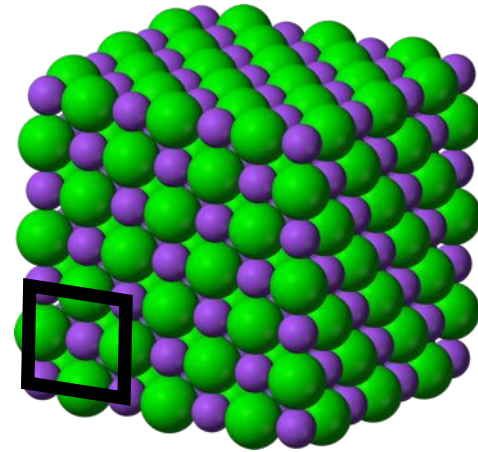
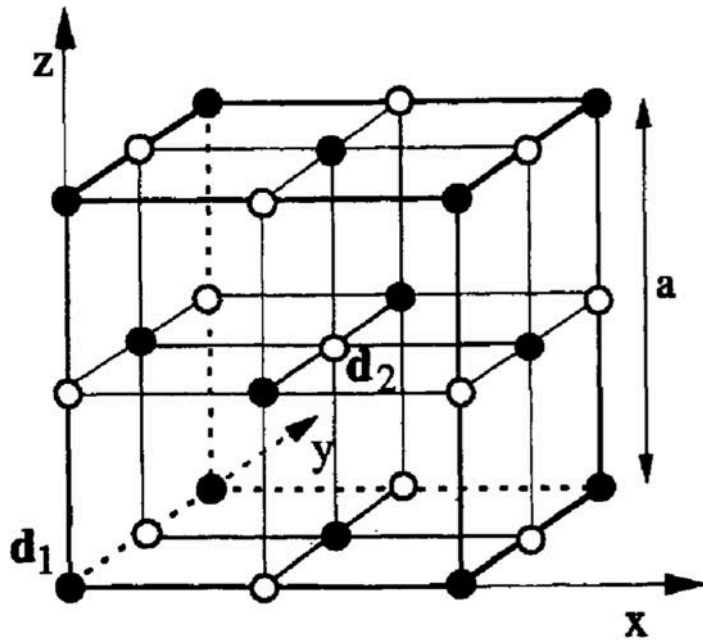
Atoms/unit cell = $(\frac{1}{8} \times 8) + (\frac{1}{2} \times 6) = 4$

Cloruro di Cesio (CsCl)



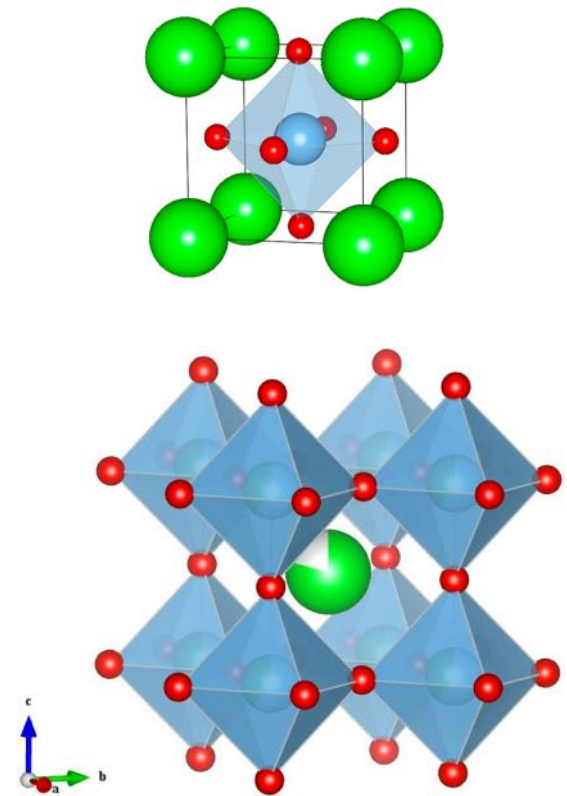
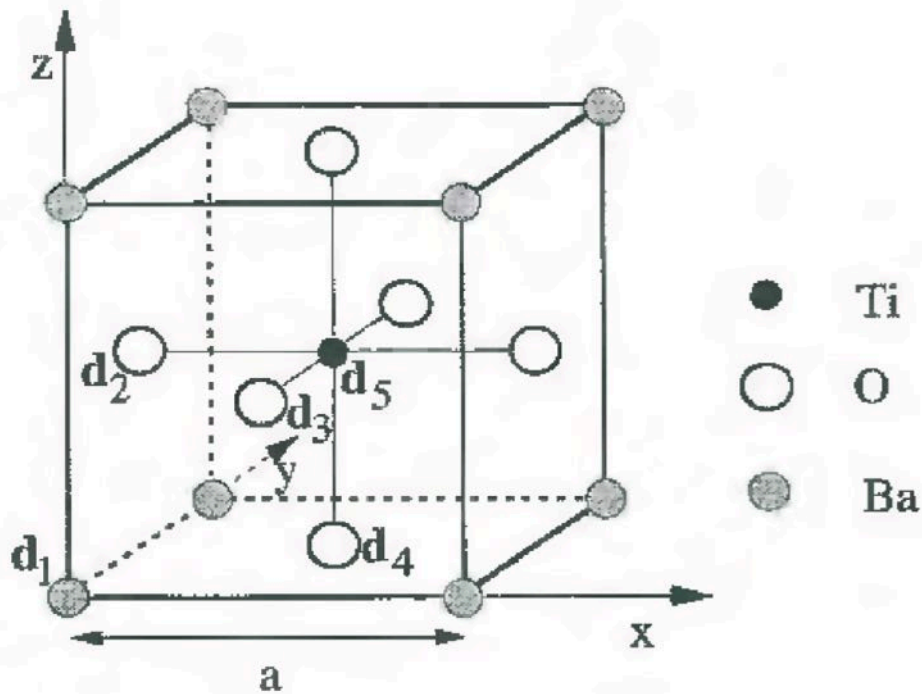
Crystal	a (Å)	Crystal	a (Å)	Crystal	a (Å)
AlNi	2.88	CsCl	4.12	TlCl	3.83
CuZn (β -brass)	2.94	CsBr	4.29	TlBr	3.97
AgMg	3.28	CsI	4.57	TlI	4.20

Cloruro di Sodio (NaCl)

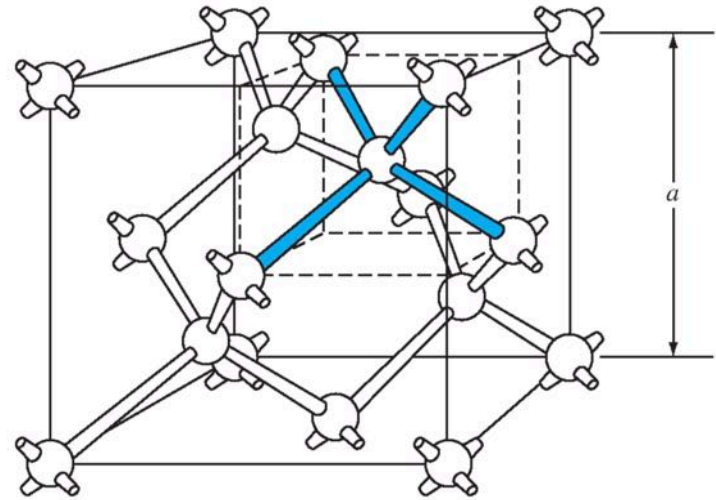
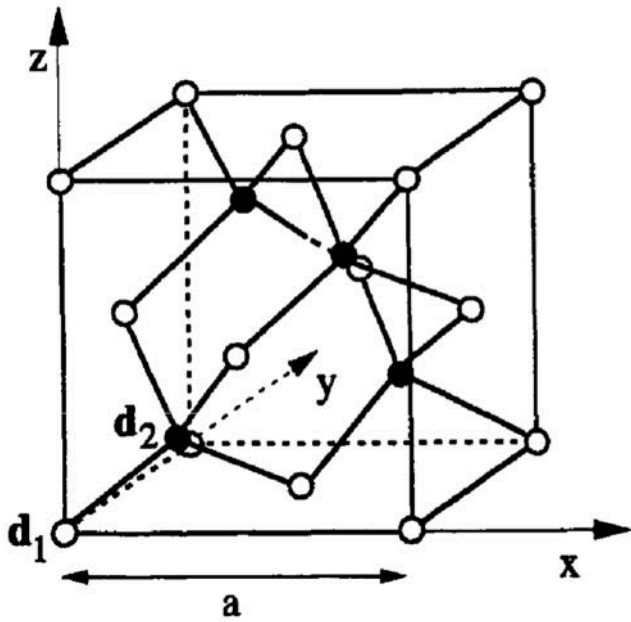


Crystal	$a(\text{\AA})$	Crystal	$a(\text{\AA})$	Crystal	$a(\text{\AA})$
LiF	4.02	KBr	6.60	MgO	4.21
LiBr	5.50	AgBr	5.77	MnO	4.43
NaCl	5.64	AgF	4.92	MgS	5.20
NaI	6.47	CaSe	5.91	PbS	5.92
KCl	6.29	BaO	5.52	SrTe	6.47

Perovskite – Titanato di Bario (BaTiO_3)

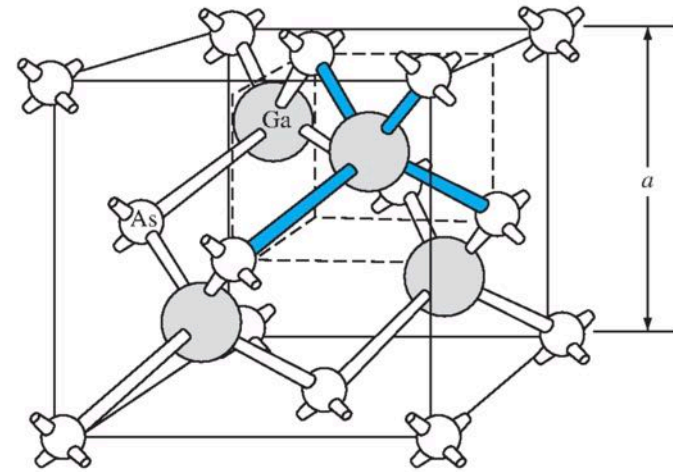
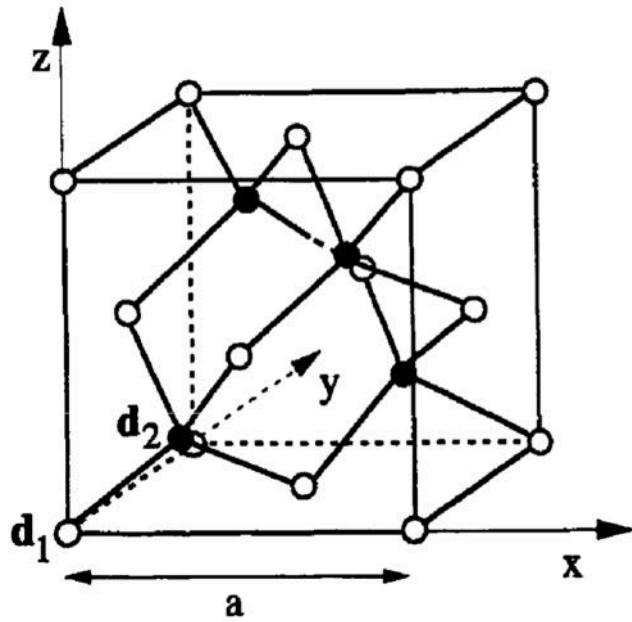


Diamante



Crystal	a (Å)
C (diamond)	3.57
Si	5.43
Ge	5.66
α -Sn (grey)	6.49

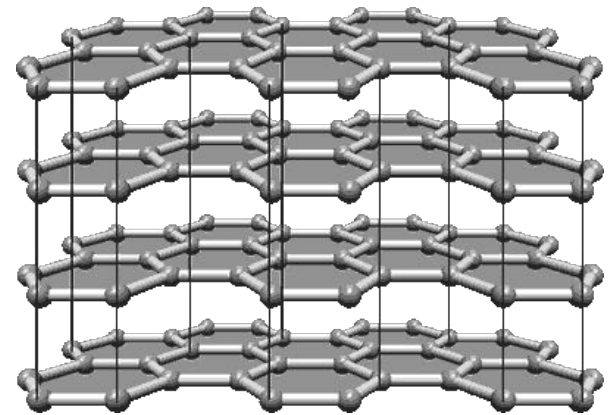
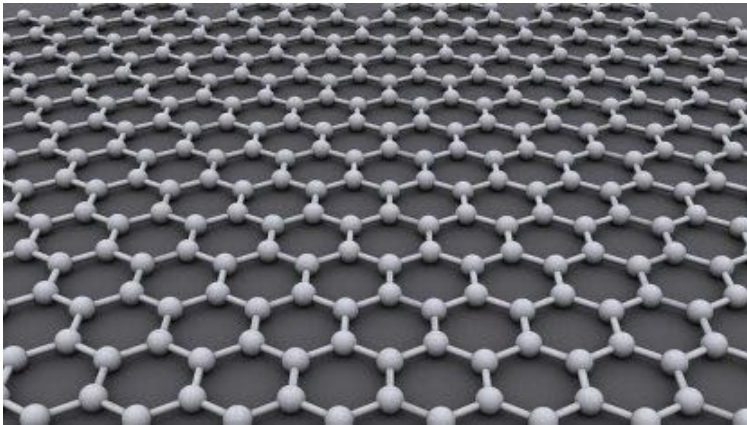
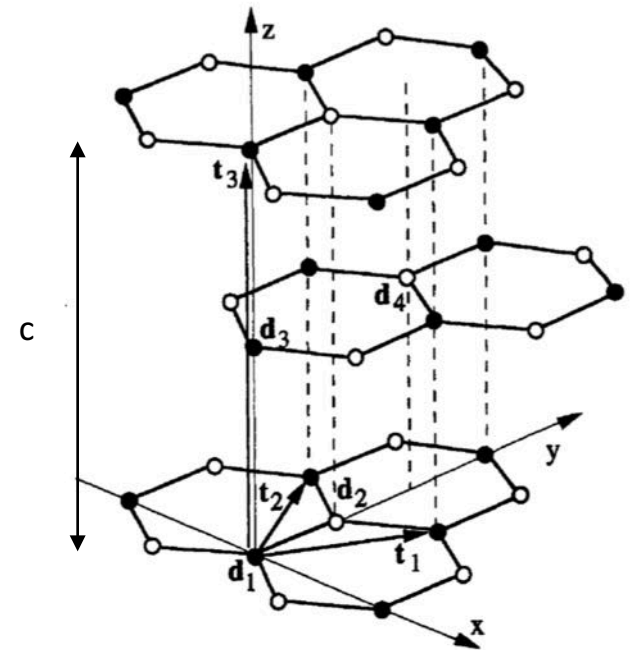
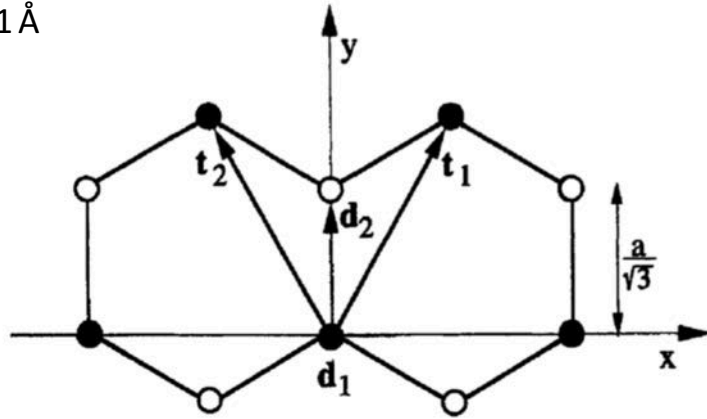
Zincoblenda



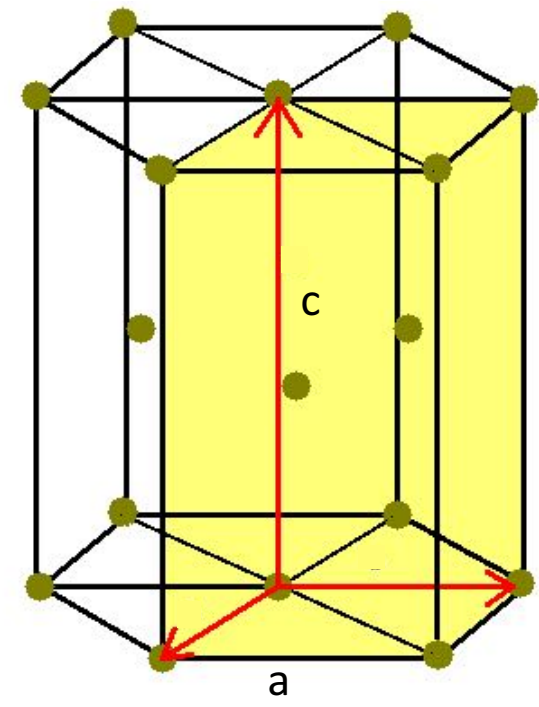
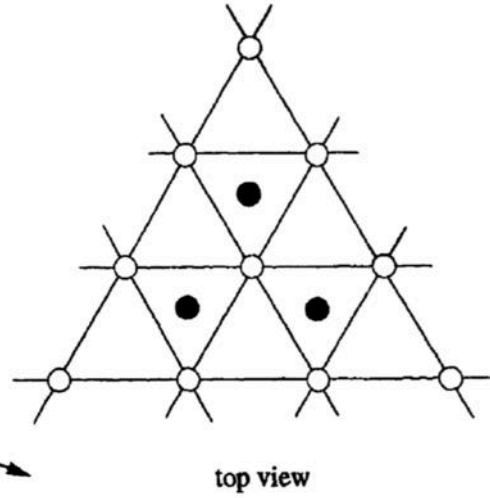
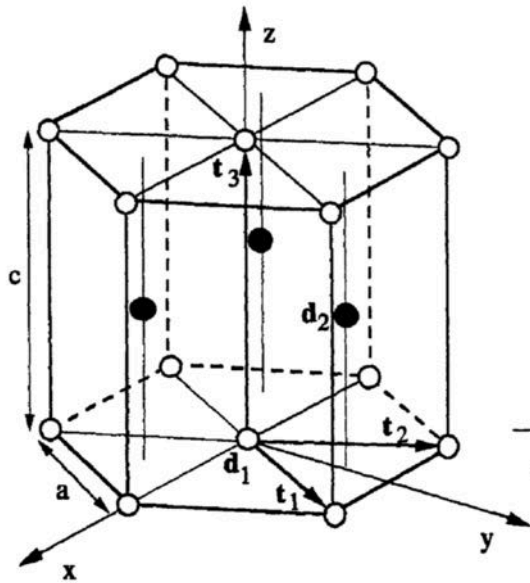
Crystal	a (Å)	Crystal	a (Å)	Crystal	a (Å)
SiC	4.35	AlP	5.45	InAs	6.04
ZnS	5.41	AlAs	5.66	InSb	6.48
ZnSe	5.67	GaAs	5.65	SiC	4.35
MnS (red)	5.60	GaSb	6.12	CuCl	5.41
CdS	5.82	GaP	5.45	CuBr	5.69
CdTe	6.48	AgI	6.47	HgSe	6.08

Grafene e Grafite

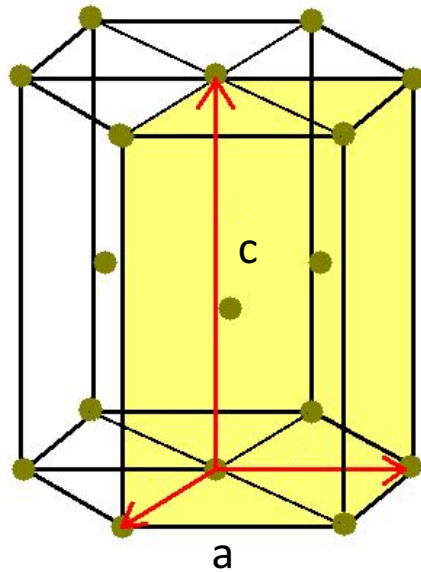
$a = 2.46 \text{ \AA}$
 $c = 6.71 \text{ \AA}$



Esagonale compatto (close-packed) - hcp



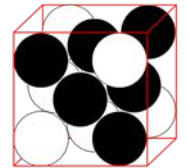
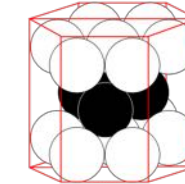
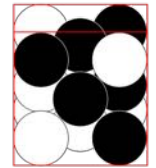
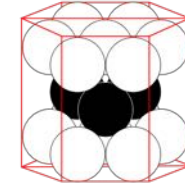
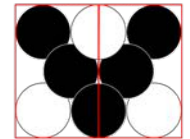
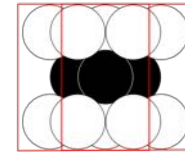
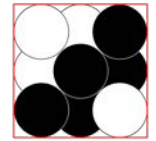
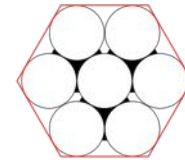
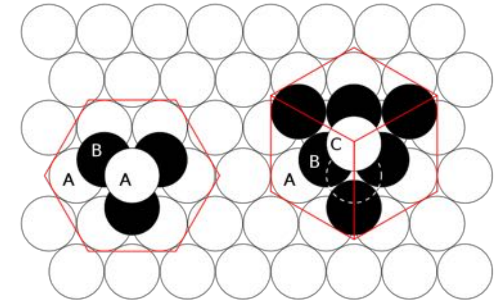
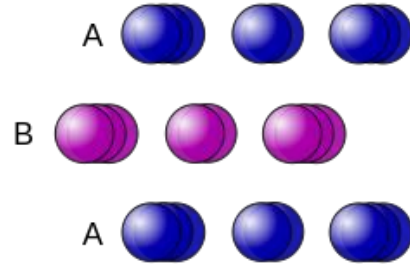
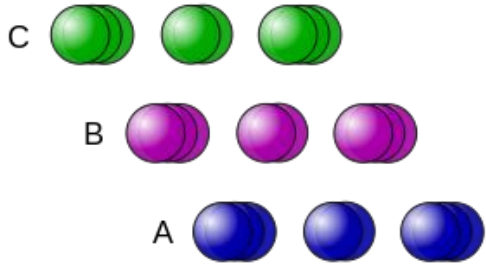
Esagonale compatto (close-packed) - hcp



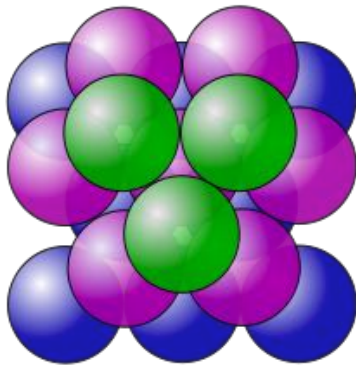
Crystal	a (Å)	c/a	Crystal	a (Å)	c/a
He	3.57	1.63	Mg	3.21	1.62
Be	2.29	1.58	Ti	2.95	1.58
Nd	3.66	1.61	Zr	3.23	1.59
Zn	2.66	1.86	Y	3.65	1.57
Cd	2.98	1.88	Gd	3.64	1.59
α -Co	2.61	1.62	Lu	3.50	1.58

$$\left(\frac{c}{a}\right)_{\text{ideale}} = \sqrt{\frac{8}{3}} \approx 1.633$$

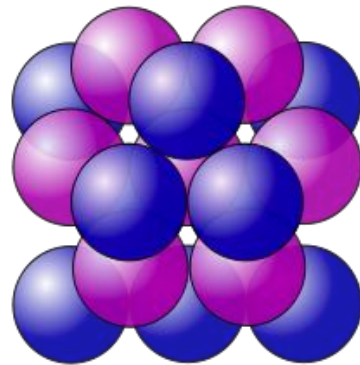
close-packed lattices



lungo (111)



Cubico a facce centrate ABC

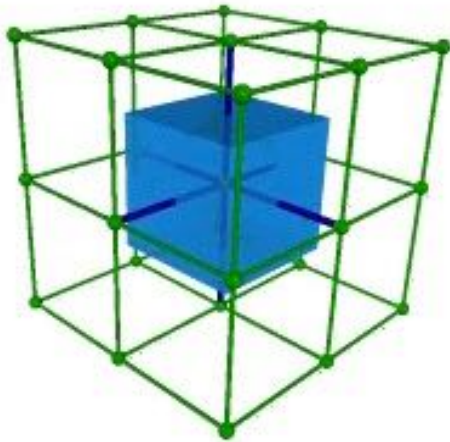


Esagonale compatto ABA

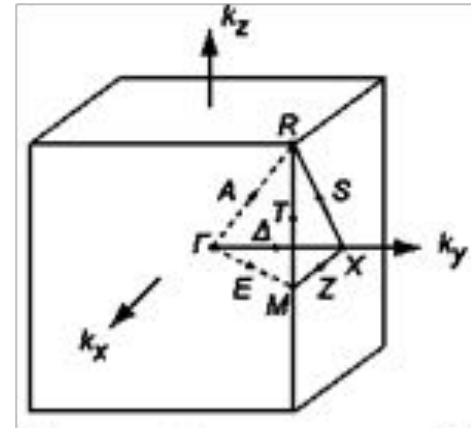
Cella di Wigner-Seitz (WS) e zona di Brillouin (BZ)

Cubico semplice

Reticolo Reale

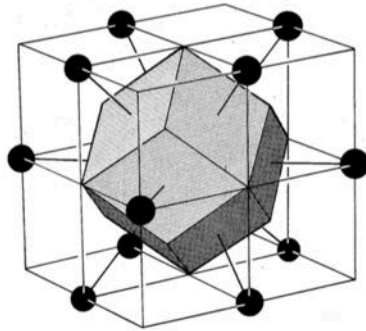


Reticolo Reciproco

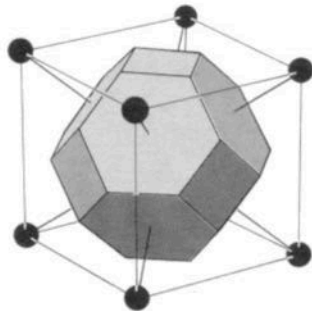


Cella di Wigner-Seitz (WS) e zona di Brillouin (BZ)

Reticolo Reale

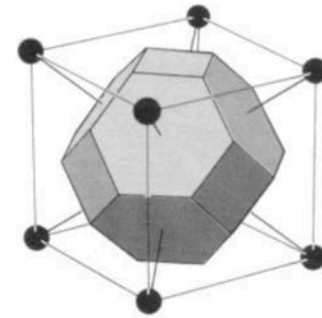


fcc WS
Dodecaedro rombico

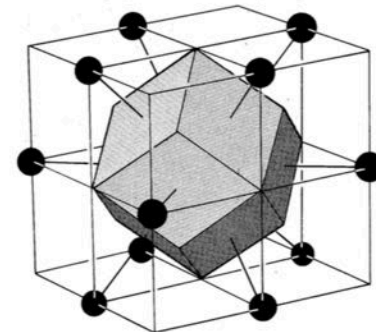


bcc WS
Ottaedro troncato

Reticolo Reciproco

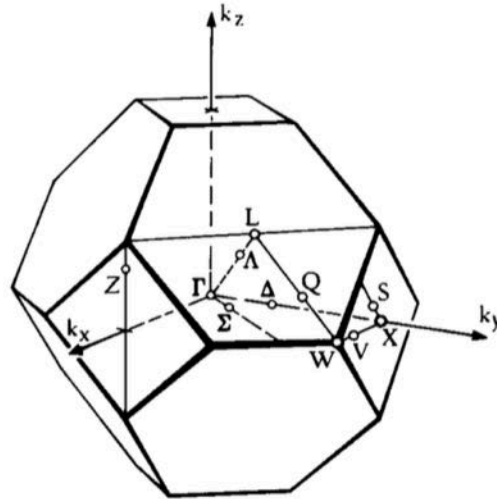


fcc BZ
Ottaedro troncato

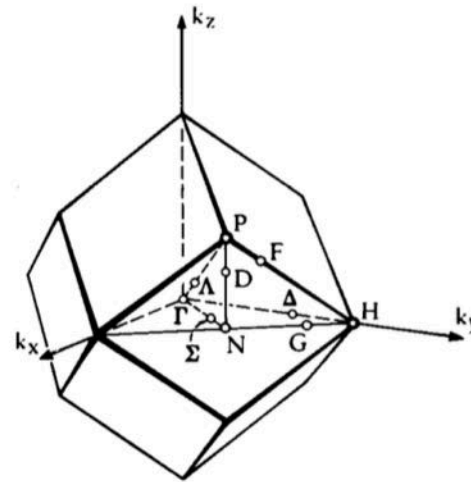


bcc BZ
Dodecaedro rombico

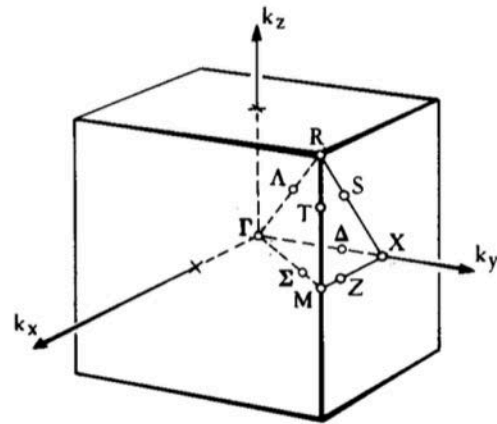
Zone di Brillouin (BZ)



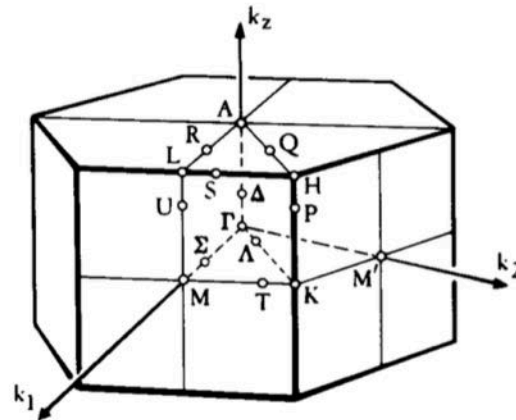
FACE CENTERED CUBIC



BODY CENTERED CUBIC



SIMPLE CUBIC



HEXAGONAL